Photosynthetic artificial organelles sustain and control ATP-dependent reactions in a protocellular system

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Inside cells, complex metabolic reactions are distributed across the modular compartments of organelles1,2. Reactions in organelles have been recapitulated in vitro by reconstituting functional protein machineries into membrane systems3–5. However, maintaining and controlling these reactions is challenging. Here we designed, built, and tested a switchable, light-harvesting organelle that provides both a sustainable energy source and a means of directing intravesicular reactions. An ATP (ATP) synthase and two photoconverters (plant-derived photosystem II and bacteria-derived proteorhodopsin) enable ATP synthesis. Independent optical activation of the two photoconverters allows dynamic control of ATP synthesis: red light facilitates and green light impedes ATP synthesis. We encapsulated the photosynthetic organelles in a giant vesicle to form a protocellular system and demonstrated optical control of two ATP-dependent reactions, carbon fixation and actin polymerization, with the latter altering outer vesicle morphology. Switchable photosynthetic organelles may enable the development of biomimetic vesicle systems with regulatory networks that exhibit homeostasis and complex cellular behaviors.

Biomimetic membrane systems have been used as bioreactors, permitting investigation of specific intracellular metabolic reactions using lipid vesicles in a simplified and isolated in vitro environment2,6,7. These approaches have been applied to study lipid–protein interactions8, membrane dynamics9,10, and cytoskeleton formation4,11,12, as well as synthesis of ATP13,14, DNA15, RNA16,17, and proteins18,19. However, this work has largely relied on spontaneous reactions, making it difficult to maintain, couple, and control cascade reactions. Photosynthesis could provide a sustainable energy source for biomimetic vesicle systems and a means of controlling coupled reactions. Converting light into energy requires at a minimum two membrane-bound proteins, a photocovertver, and an ATP synthase5. The photoconverter establishes an electrochemical gradient across the membrane, generating a proton motive force (PMF) that drives conversion of ADP to ATP by ATP synthase. ATP synthesis would provide the energy needed for ATP-dependent intravesicular metabolic reactions. As photoconverters from different species have diverse biophysical properties20, it should be possible to directly control PMF generation and ATP synthesis by selecting and combining photoconverters with specific properties. Pioneering studies have demonstrated that reconstituting a photoconverter and an ATP synthase into proteoliposomes enables ATP synthesis13,14. However, these proteoliposomes did not dynamically control chemical reactions or couple with other intravesicular reactions.

We engineered switchable photosynthetic organelles (~100 nm diameter) as energy modules and encapsulated them within a giant lipid vesicle (~10–100 µm diameter) embedded with ionophores to form an encapsulated organelle system (Fig. 1). Activation of the two photoconverters, plant-derived photosystem II (PSII) and bacteria-derived proteorhodopsin (PR), generates PMF across the artificial organelle’s membrane, inducing the ATP synthase to catalyze the conversion of ADP to ATP5 (Fig. 1a). We selected PSII and PR as photoconverters owing to their unique photophysical properties20. PSII is a large protein complex (350 kDa)21 containing 35 chlorophyll molecules inside chromophores, and PR is a single protein (27 kDa)22 holding only a single retinal chromophore. Blue and red light activates electron transport chains in the PSII reaction center21, whereas mainly green light initiates direct proton pumping in PR22. Moreover, PR exhibits pH-dependent bidirectional proton-pumping ability, determined by the pK_a of the primary proton acceptor residue (D97)23. At a higher pH, both PR and PSII work in conjunction to increase the PMF, but at lower pH, PR changes its pumping direction, counteracting the action of PSII. Thus, ATP synthesis can be dynamically facilitated or impeded by stimulation of PSII and PR with red or green light, respectively (Fig. 1b). Finally, we coupled the switchable organelle photosynthesis with actin polymerization, enabling us to control actin assembly inside the giant lipid vesicle and the dynamic interaction between actin filament growth and the membrane (Fig. 1c).

Received 5 April 2017; accepted 11 April 2018; published online 28 May 2018; doi:10.1038/nbt.4140
We assembled and tested the encapsulated organelle system progressively: first, the proteoliposome (reconstitution of photoconverters into vesicles), then the photosynthetic organelle (reconstitution of ATP synthase into proteoliposomes), and, finally, the photosynthetic protocell system (encapsulating the photosynthetic organelles into a giant lipid vesicle; see Supplementary Fig. 1 for detailed information about lipid compositions and ionic solutions used during testing).

We purified recombinantly expressed PR22 from *Gamma proteobacterium* and PSII from *Spinacia oleracea* and reconstituted the purified proteins into vesicles (Fig. 2a, and Supplementary Fig. 2, and Online Methods). The coexistence of the two proteins in the PSII–PR co-reconstituted proteoliposomes was demonstrated by their black color, resulting from the summation of the individual absorbance bands of PSII (440 nm and 670 nm peaks) and PR (520 nm peak) (Fig. 2b). Upon white light illumination, the PSII–PR co-reconstituted proteoliposomes generated PMF across their membrane. Ratiometric pH measurement with an encapsulated pH indicator (SNARF-1) (Supplementary Figs. 3 and 4) and patch clamp recording revealed that the two photoconverters in the lipid bilayers that formed on planar patch clamp glass chips produced photocurrent (Fig. 2c,d), which led to a change in the intravesicular pH of the proteoliposomes (Fig. 2e,f and Supplementary Fig. 3e). The photosensitive current and the change in the intravesicular pH of the PSII–PR co-reconstituted proteoliposomes were in ranges similar to the estimated values as calculated from measurements of single PSII or PR reconstituted proteoliposomes (Fig. 2d,f), indicating robust performance of the two photoconverters in the lipid membrane. In addition, the pH gradient was sharply decreased by removing a negatively charged phospholipid, 1-palmitoyl-2-oleoyl-sn-glycero-3-phospho-(1'-rac-glycerol) (2.4 × for PR, 2.9 × for PSII), which confirmed that the negatively charged phospholipid improved the orientation and activity of PSII and PSII24,25 (Supplementary Fig. 5).

Next, ATP synthase purified from *Bacillus pseudofirmus* was reconstituted into PSII–PR co-reconstituted proteoliposomes. The PMF driven by optical activation was sufficient to induce ATP synthesis. The maximum turnover number for ATP synthase was 4.3 ± 0.1 s⁻¹, measured by the luciferin–luciferase assay. This indicates that the ATP synthesis per ATP synthase achieved was up to 13.3–32.3% of the ATP turnover number of *Escherichia coli* (13–33 s⁻¹)26 (Supplementary Note 1). ATP was produced outside the artificial organelles at the sites where ATP synthases were integrated into the proteoliposomes (Fig. 2g). ATP synthesis sharply decreased upon treatment with a protonophore (carbonyl cyanide-p-trifluoromethoxyphenylhydrazone) and an inhibitor of ATP synthase (oligomycin), demonstrating that ATP synthesis was directly coupled with the light-driven electrochemical potential gradient (Fig. 2g). The artificial organelle sustained ATP conversion for 3 days (half-maximum efficiency) at room temperature and 1 month at 4 °C (Fig. 2h).

**Figure 1** Design and applications of the photosynthetic artificial organelle. (a) Upon optical stimulation, the artificial organelle synthesizes ATP by the coordinated activation of two complementary photoconverters (PSII and PR) and an ATP synthase reconstituted into the organelle’s membrane. (b) Activation of PSII with red light facilitates ATP synthesis by generating protons inside the organelle, while activation of PR with green light impedes ATP synthesis by depleting protons. (c) After organelle energy modules are encapsulated in a giant vesicle, optical stimulation couples ATP synthesis with ATP-dependent actin polymerization and morphological change of the vesicle.
Figure 2. Light-driven ATP synthesis in the artificial organelle. (a) Suspension of small proteoliposomes and fluorescence images of small proteoliposomes and giant proteoliposomes containing the photoconverters PSII and PR. Similar results were obtained from three independent experiments, and representative images are shown. (b) Absorption spectrum of the proteoliposomes, demonstrating the reconstitution of PSII and PR into vesicles. The number of chromophores in the proteoliposomes (PSII and PR) was calculated by the absorption spectrum and absorption coefficients of the chromophores (Supplementary Fig. 2). Data shown represent seven independent experiments with similar results. (c–f) Upon white light stimulation, PSII and PR elicited photocurrents (c, d, n = 7 biologically independent experiments), leading to changes in intravesicular pH (e, f, n = 7 biologically independent experiments). There was no difference between the measured photosensitive current and pH gradient generation of PSII–PR co-reconstituted proteoliposomes and the estimated values of these measurements as calculated from measurements of single-photoconverter-reconstituted proteoliposomes, showing independent activity of photoconverters in vesicles (d, f). (g) Upon white light stimulation, reconstituted ATP synthase catalyzed the synthesis of ATP. ATP synthesis was driven by a light-induced pH gradient across the membrane, as shown by significantly decreased ATP synthesis after treatment with a protonophore (FCCP) and an inhibitor of ATP synthase (oligomycin). P = 3.5 × 10⁻⁵ (PR vs. oligomycin at 20 min). ATP concentration was measured with luciferin–luciferase assay (n = 7 biologically independent experiments). (h) Photosynthetic activity of the artificial organelle over 1 month (n = 3 biologically independent samples). (i) Coupling of ATP synthesis with carbon fixation. Optical illumination increased the concentration of OAA, whereas phenylacetic acid, a pyruvate carboxylase inhibitor, suppressed it. OAA concentration was measured with fluorometric assay (n = 7 biologically independent experiments). (j) Gas chromatography/mass spectrometry (GC/MS) analysis of carbon fixation products. Upper panel: GC/MS peak ratios (Supplementary Fig. 6) of chromatograms between pyruvate and OAA were calculated as a function of illumination time. Lower panel: the mass spectroscopy confirmed the presence of OAA. Data shown represent two independent experiments with similar results. (d–f, g) Error bars and the colored bands indicate mean ± s.d. of independent experiments. (d, f) P values were calculated by Student’s two-sided t-test.

We tested whether the synthesized ATP could be used as an energy source for the coupled reaction of carbon fixation (Fig. 2i, j and Supplementary Fig. 6). Both fluorometric assay and gas chromatography–mass spectrometry (GC/MS) showed that the four-carbon compound oxaloacetate (OAA) was formed from the three-carbon compound pyruvate in the presence of pyruvate carboxylase, acetyl-coenzyme A, and bicarbonate, which means that the synthesized ATP was converted back into ADP and reused in the ATP–ADP cycle.
The conversion of pyruvate into OAA was abolished by phenylacetic acid, an inhibitor of pyruvate carboxylase activity, indicating direct coupling of carbon fixation with ATP synthesis (Fig. 2i and Supplementary Fig. 7).

PMF generation by the photosynthetic organelle could be controlled through the wavelength- and pH-dependent activities of the photoconverters (Supplementary Video 1). PSII–PR co-reconstituted proteoliposomes generated wavelength- and pH-dependent photocurrents (Fig. 3a). Consistent with their absorbance peaks (Fig. 2b), PSII and PR could be independently activated with red and green light, respectively, with minimal cross-talk (Fig. 3b). PSII activation generated a negative current at both pH 6.5 and pH 8, with significantly reduced magnitude at pH 6.5, whereas PR activation generated a negative current at pH 8.0 but a positive current at pH 6.5 (Fig. 3b and Supplementary Figs. 8 and 9). The wavelength- and pH-dependent activity of the photoconverters set the direction of change in intravesicular pH (Fig. 3c). The intravesicular pH of PSII reconstituted proteoliposomes decreased with a gradual slope, while the pH-sensitive bidirectionality of PR pumping decreased the intravesicular pH of PR reconstituted proteoliposomes at pH > 7.4 but increased it at pH < 7.4 (Supplementary Fig. 10). The pH dependence of PR determined the bidirectional intravesicular pH range (pH > 7.4 of PSII–PR co-reconstituted proteoliposomes; at the bidirectional pH range, intravesicular pH was decreased by independently activating PSII with red light and increased by activating PR with green light (Fig. 3c and Supplementary Video 1). In contrast, at pH > 7.4, illumination led to only a decrease in intravesicular pH, determining the upper limit of intravesicular pH as 7.4 (Fig. 3c).

Switchable PMF generation by the photosynthetic organelle enabled control of ATP synthesis (Fig. 3d,e and Supplementary Fig. 11). The density ratios of PSII, PR, and ATP synthase over lipid (19,000:1, 260:1, and 2,000:1, respectively) were chosen to ensure that PSII and PR would contribute similarly to the organelles’ intravesicular pH level and to maximize ATP synthesis (Supplementary Fig. 12). The initial pH level (>7.8) was selected to keep the pH level outside the organelles higher than the upper limit of intravesicular pH (7.4), thus maintaining a positive PMF for ATP synthesis. White light illumination for 15 min was enough to bring the intravesicular pH to the bidirectional pH range (Fig. 3d,e). As expected, ATP synthesis was facilitated by independently activating PSII with red light and impeded by activating PR with green light when intravesicular pH was within the bidirectional pH range (Fig. 3d). Thus, PR activation improved the quenching rate of ATP synthesis compared to the light-off case (Fig. 3e and Supplementary Fig. 13) because the outward photosensitive current of PR rapidly depleted protons inside the organelle (Fig. 3c). This wavelength- and pH-dependent control of ATP synthesis via independent activation of PSII and PR provided a mechanism for controlling coupling reactions, such as carbon fixation (Fig. 3f and Supplementary Fig. 14).

Finally, we applied the artificial organelles to simulate a ubiquitous process—cytoskeleton formation through ATP-dependent actin polymerization27. We embedded magnesium ionophores in a giant vesicle and encapsulated the organelles and the reactants (ADP and G-actin) within it to form the photosynthetic protocellular system (Supplementary Fig. 15). To improve encapsulation efficiency, we optimized the artificial organelles’ size and electrical charge (Supplementary Fig. 16). The initial pH inside the protocellular system was maintained at 7.8, and magnesium ions were provided outside it. Optical stimulation initiated ATP synthesis and induced ATP-dependent actin polymerization, leading to growth of three-dimensional actin filaments (Fig. 4a, Supplementary Fig. 17, and Supplementary Video 2). The ATP triggered G-actin nucleation, and then actin filaments elongated and tangled to form an F-actin sphere that occupied 75–100% of the interior of the protocellular system. Eventually the actin sphere grew large enough to rupture the
membrane (Fig. 4b and Supplementary Video 3). Red light facilitated and green light impeded filament growth via independent PSII and PR activation (Fig. 4c and Supplementary Video 4). In addition, PR activation by green light improved the quenching rate of actin polymerization compared with the light-off case (Fig. 4d and Supplementary Video 5).

Both our analytical model and our experimental results demonstrated that ATP synthesis by the artificial organelles was the dominant process in actin filament growth. In the analytical model, the increase of the radius of the F-actin sphere was predicted to be proportional to the amount of ATP produced over time, that is, \( R(t) \propto ATP(t) \), when the growth-rate-limiting process is associated with diffusion of synthesized ATP from the organelles to the existing F-actin sphere. Figures 2g and 3d show white-light-induced ATP production that follows the relation \( ATP \propto t^{1/2} \), while Figure 4b shows that the projected area of the F-actin sphere grows linearly in time, that is, \( R^2(t) \propto t \), which is consistent with the argument that ATP diffusion is the limiting process. These observations yield \( R(t) \propto ATP(t) \) (Supplementary Fig. 18 and Supplementary Note 1). In addition, actin filaments did not form in the absence of light stimulation, artificial organelles, or ADP, indicating that actin polymerization was completely dependent on ATP synthesized by the artificial organelles (Supplementary Fig. 19 and Supplementary Videos 6–9).

Moreover, cytoskeleton formation and manipulation induced morphological change of the outer membrane of the protocellular system via membrane–actin interactions27,28. As actin filaments grew, they protruded into the outer membrane, deforming and eventually rupturing it (Fig. 4b and Supplementary Videos 3 and 10–12). This behavior confirmed that the elastic interaction between the actin filaments and the membrane was the sole cause of membrane deformation, without additional bundling or the formation of tip-complex proteins.11,12 To control membrane–actin interactions, we used three different membrane mixtures exhibiting a range of attraction–repulsion interactions with actin filaments: a liquid-disordered (Ld) or a liquid-ordered (Lo2) phase membrane. Polymerized actin was strongly attached to the Ld phase membrane (e) but was separated from the Lo2 membrane by a gap (f). (g) Strong attractive interactions between the Ld phase membrane and actin filaments deformed the local membrane, leading to a broad distribution of local curvature, while the inhibition of membrane–actin interactions between the Lo2 phase membrane and actin filaments maintained a circular membrane shape with a narrow distribution of local curvature. (h) The global deformation of phase-separated membranes composed of Ld and Lo mixtures. (i,j) Actin polymerization deformed the spherical membrane system into teardrop (i) or mushroom (j) shapes via heterogeneous membrane–actin interactions. (e–j) The data and images are representative of three independent experiments with similar results.

Figure 4 Control of actin polymerization and membrane shape of the photosynthetic protocellular system by light. (a) Three-dimensional reconstruction of the complete system from confocal microscopy images. The ATP-dependent polymerization of actin filaments (white lines) was triggered by red light activation of the artificial organelles. Similar results were obtained from two independent experiments, and representative images are shown. (b) Optical stimulation initiated G-actin nucleation (~15 min) and F-actin elongation (~60 min), leading to formation of a single F-actin sphere (~135 min). The F-actin sphere grew until the membrane ruptured (260 min). Similar results were obtained from three independent experiments, and representative images and data are shown. (c) Actin polymerization was controlled by red and green light. The data are representative of four independent experiments with similar results. (d) Optical control of actin growth (\( n = 4 \) biologically independent experiments). Error bars indicate mean \( \pm \) s.d. of four independent experiments. \( P \) values were calculated by Student’s two-sided \( t \)-test. (e–j) Membrane deformation caused by impingement of polymerized actin on the outer membrane. (e,f) Local deformation of a liquid-disordered (Ld) or a liquid-ordered (Lo2) phase membrane. Polymerized actin was strongly attached to the Ld phase membrane (e) but was separated from the Lo2 membrane by a gap (f). (g) Strong attractive interactions between the Ld phase membrane and actin filaments deformed the local membrane, leading to a broad distribution of local curvature, while the inhibition of membrane–actin interactions between the Lo2 phase membrane and actin filaments maintained a circular membrane shape with a narrow distribution of local curvature. (h) The global deformation of phase-separated membranes composed of Ld and Lo mixtures. (i,j) Actin polymerization deformed the spherical membrane system into teardrop (i) or mushroom (j) shapes via heterogeneous membrane–actin interactions. (e–j) The data and images are representative of three independent experiments with similar results.
phase mixture (Lo2) composed of sphingomyelin, cholesterol, and polysaturated 1,2-dioleoyl-sn-glycero-3-phosphoethanolamine-N-[methoxy(polyethylene glycol)-2000] (polysaturated-PEG2000-PE) that produced a strong repulsive interaction (Supplementary Fig. 20). Upon optical stimulation, the single-phase membrane system made of Lo and Lo1 altered local membrane curvature owing to the attractive interaction, but the system made of Lo2 maintained a spherical shape owing to the repulsive interaction (Fig. 4e–g, Supplementary Fig. 20, and Supplementary Videos 10–12). In the phase-separated membrane system made of two different mixtures (such as Ld−Lo1 and Ld−Lo2) (Fig. 4h and Supplementary Fig. 21), the difference between local membrane–actin interactions induced changes in the global cell curvature, deforming the spherical cells into teardrop or mushroom shapes (Fig. 4i,j, Supplementary Figs. 21 and 22, and Supplementary Video 13).

Our demonstration that switchable photosynthetic organelles can control ATP-dependent cytoskeleton formation and thus the morphology of the protocellular system supports the possibility of developing artificial cell-like systems that replicate homeostasis and complex cellular behaviors, in vitro. Just like chloroplasts and mitochondria in nature, the photosynthetic artificial organelle presented here served as an energy module for the protocellular system. The resulting system was able to induce and control cascading metabolic reactions (carbon fixation, actin polymerization) as well as exhibit cell-like behavior (morphological change). We envision that our bottom-up approach could be used to create isolated modules representing additional organelles, such as the endoplasmic reticulum for protein metabolism or nucleus for genetic replication. Introducing networks of proteins and organelles into artificial cell-like environments may contribute to achieving the long-standing goal of building a cell de novo29,30.

METHODS

Methods, including statements of data availability and any associated accession codes and references, are available in the online version of the paper.

Note: Any Supplementary Information and Source Data files are available in the online version of the paper.

ACKNOWLEDGMENTS

This work was supported by the Mid-Career Researcher Programs (2016R1A2B3015239 and 2011-0017539), Foreign Research Institute Recruitment Program (2013K1A1A3055268, 2015R1D1A1A01058917, and DRC-14-03-KRICT through the National Research Foundation, funded by the Ministry of Science and ICT, Korea. K.Y.L. and T.K.A. acknowledge the support provided by the Woo Jang Chun Special Project of the Rural Development Administration, Korea (PJ009106020213).

AUTHOR CONTRIBUTIONS


COMPETING INTERESTS

The author declares no competing interests.

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Photosystem II (PSII) was purified from 

Overnight cultures of transformed cells in UT5600 were diluted 1:100 and induced to the dopsin (MBP) pKJ900 carrying the proteorhodopsin (PR) gene (Monterey Bay proteorhodopsin) on sucrose gradients in a Beckman Coulter SW 28 rotor (Beckman Coulter, Brea, CA) at −80 °C for 4 h. Thawing was complete. The solubilized samples were centrifuged at 80,000 rpm for 1 min. The tubes were then kept in a 4 °C refrigerator until thawing. The sucrose density gradients prepared by freezing and thawing were resuspended in a small volume of 10 mM HEPES (pH 7.5), 0.4 M sorbitol, and 5 mM MgCl₂, and thawed. The sucrose gradient method (Sigma–Aldrich Co. (St. Louis, MO)). Phenyl-β-d-maltopyranoside (D MDMA) was obtained from Anacetra, Inc. (Maumee, OH). Polyoxymethylene Bio-Beads SM-2 (25–50 mesh) was purchased from Bio-Rad, Inc. (Hercule, CA). The fluorescent probes Alexa Fluor 488 TFP ester (alexa-488) and 3-(and-6-carboxy SNARF-1 (SNARF-1) and an ATP Determination Kit were purchased from Molecular Probes (Eugene, OR). Phenyl-β-naproximone was purchased from Acros Organics (Morris Plains, NJ). An Oxaalocetase Assay Kit and other chemicals and reagents were purchased from BioVision, Inc. (Mountain View, CA). Actin protein and Actin-stabilized 670 phallolidin were purchased from Cytoskeleton, Inc. (Denver, CO). Carboxyl cyanide (3-trifluoromethoxy)phenyl hydrazide (FFCP), N-Methyl-N-(tert-butylidimethylsilyl)trifluoroaceticamide (MTBSTFA), with 1% Tert-Butylmethylthylchlorosilane (1% TBDMS) and isopropyl β-d-1-thiogalactpyranoside (ITPG) were purchased from Sigma–Aldrich Co. (St. Louis, MO).

Purification of photosystem II. Photosystem II (PSII) was purified from spinach using the sucrose gradient method. Leaves were gently ground in a buffer solution containing 20 mM Tricine-KOH (pH 7.8), 0.4 M NaCl, 5 mM MgCl₂, and the protease inhibitors benzamidine (0.2 mM), h-aminocaproic acid (1 mM), and phenylmethylsulfonyl fluoride (PMSF) (0.2 mM). The solution was centrifuged for 10 min at 1,000 rpm, and the pellet was resuspended in a buffer solution containing 20 mM Tricine-KOH (pH 7.8), 0.15 M NaCl, 5 mM MgCl₂, 0.2 mM benzamidine, 1 mM h-aminocaproic acid, and 0.2 mM PMSF. This solution was centrifuged for 10 min at 40,000 rpm, and the supernatant was resuspended in 20 mM HEPES (pH 7.5), 15 mM NaCl, and 5 mM MgCl₂, and the solution was centrifuged again for 10 min at 6,000 rpm. To solubilize stacked thylakoids, the mixture was incubated in 3% DDM (w/v), 15 mM NaCl, and 5 mM MgCl₂ for 30 min on ice with soft agitation. To remove nonsolubilized material, a 5-min centrifugation at 2,000 rpm was performed. Membranes were resuspended in a small volume of 10 mM HEPES (pH 7.5), 0.4 M sorbitol, 10 mM NaCl, and 5 mM MgCl₂. Sucrose gradients were prepared by freezing and thawing. The sucrose density gradients prepared by freezing and thawing were made with 0.7 M sucrose, 10 mM HEPES (pH 7.5), 10 mM NaCl, 5 mM MgCl₂, and 0.03% (w/v) DDM. These were mixed and kept in thin-wall Ultra-Clear tubes (Beckman Coulter, Brea, CA) at −80 °C for 4 h. Thawing of the frozen mixture was initiated by immersing the tubes in water at room temperature for 1 min. The tubes were then kept in a 4 °C refrigerator until thawing was complete. The solubilized samples were centrifuged at 80,000 rpm on sucrose gradients in a Beckman Coulter SW 28 rotor (Beckman Coulter, Brea, CA) for 40 h at 4 °C. We were able to detect eight distinct bands (B0–B7) containing PSII core (Supplementary Fig. 23a). The protein composition of each band was determined by (SDS-PAGE) and an absorption spectrum was obtained to confirm the condition of PSII (Supplementary Fig. 23c).

Heterologous expression and purification of proteorhodospin. The plasmid pK3900 carrying the proteorhodospin (PR) gene (Monte Bay proteorhodospin (MBP) Gamma proteobacterium, database ID: AAG10475) was introduced to the Escherichia coli UT5600 strain for heterologous expression. Overnight cultures of transformed cells in UT5600 were diluted 1:100 and grown to produce an absorbance of 0.4 OD units at 600 nm (0.4 OD600) at 30 °C. Cells were induced with 1 mM IPTG and 10 μM all-trans-Retinal (Sigma–Aldrich Co., St. Louis, MO) at 0.4 OD600 and incubated for 4 h at 30 °C. Induced cells were harvested by centrifugation at 2,500 and 4 °C for 20 min (Eppendorf centrifuge 5810R, Germany) and suspended in 50 mM Tris-HCl (pH 7.0) and 150 mM NaCl. The collected cells were sonicated (Branson Sonifier 250, Danbury, CT) for 10 min on ice and centrifuged again at 2,500g and 4 °C for 20 min to discard unbroken cells. The soluble fraction was taken and ultracentrifuged (Beckman XL-100 ultracentrifuge, rotor type 70Ti, Beckman Coulter, Indianapolis, IN) at 35,000 rpm and 4 °C for 1 h to isolate the membrane fraction. The suspended pellet was treated with 1% DDM for 1 h at 4 °C. The solubilized fraction was incubated with Ni-nitrotriacetic acid (NTA)-agarose beads (Qiagen, Venlo, Netherlands), washed with buffer (50 mM Tris-HCl (pH 7.0), 150 mM NaCl, 0.02% DDM, and 20 mM imidazole (Sigma–Aldrich Co., St. Louis, MO)). Finally, protein was eluted with buffer (50 mM Tris-HCl (pH 7.0), 150 mM NaCl, 0.02% DDM, and 250 mM imidazole). The colored fraction was collected and concentrated with an Amicon Ultra-4 10,000 MWCO centrifugal filter (Millipore Corp., Bedford, MA). The purified proteins were finally solubilized in 50 mM Tris-HCl (pH 7.0), 150 mM NaCl, and 0.02% DDM. Purified PR was subjected to SDS-PAGE to check its approximate purity (Supplementary Fig. 24a).

Purification of ATP synthase. Overall purification procedures were described previously and suitably modified. Bacillus pseudofirmus (ATCC 700159, ATCC, Manassas, VA) was grown in an alkaline nutrient medium with 0.5% NaCl at pH 9.7. Cells were grown to late log phase and then harvested by centrifugation. Cells were resuspended in 50 mM Tricine-KOH (pH 8.0), 200 mM KCl, 5 mM MgCl₂, D Nase I, and RNase A (0.1 mg/g of cell; Sigma–Aldrich Co.) were added before incubation at room temperature for 15 min with gentle rocking. All subsequent steps were performed at 4 °C. Pellets were collected by centrifugation at 125,000 rpm for 1 h. The pellets were resuspended with 160 ml of 50 mM Tricine-KOH (pH 8.0), 200 mM KCl, and 1× protease inhibitor cocktail (GenDEPOT Inc., Barker, TX) and incubated at 37 °C for 30 min with gentle rocking. 15 mM MgCl₂, D Nase I, and RNase A (0.1 mg/g of cell; Sigma–Aldrich Co.) were added before incubation at room temperature for 15 min with gentle rocking. All subsequent steps were performed at 4 °C. Pellets were collected by centrifugation at 125,000 rpm for 1 h. The washed membrane vesicles were suspended to about 10–15 mg/m in 20% glycerol, 50 mM Tricine-KOH (pH 8.0), 25 mM K₂SO₄, and 5 mM MgCl₂. Stock solutions of 0.5 M DDM and 50 mg/ml asolectin (50 mg/ml stock solution in 50 mM Tricine-KOH (pH 8.0), 25 mM K₂SO₄, 5 mM MgSO₄, 50 mM octyl glucoside, and 1 mM dithiothreitol) were added to final concentrations of 50 mM and 3 mg/ml. The suspension was incubated on ice for 30 min with occasional shaking and centrifuged at 235,000 rpm for 2 h. The grayish brown pellet was discarded, and the dark orange supernatant that contained most of the ATP synthase was subjected to ammonium sulfate fractionation after the addition of sodium chloride to 1% (100 mg/10 ml). Ammonium sulfate powder was added to the crude extract to 30% saturation (175 mg/ml) with stirring at 4 °C. The suspension was centrifuged at 75,000 rpm for 15 min, and the resulting pellets, which contained negligible ATP synthase activity, were discarded. The supernatant was brought to 60% saturation (198 mg/ml) with stirring at 4 °C. This was centrifuged at 75,000 rpm for 15 min to separate the supernatant from the ATP synthase–containing pellet. The supernatants were removed, and the pellets were dissolved in 5% glycerol, 50 mM Tricine-KOH (pH 8), 5 mM MgSO₄, and 1 mM dithiothreitol to 20–30 mg/ml. Portions (0.2 ml) were loaded onto 1–40% sucrose gradients. The gradient volume was 4.75 ml, and the tubes were centrifuged for 17 h in a Beckman-Coulter SW 50.1 rotor (Beckman Coulter, Brea, CA) at 200,000 rpm at 4 °C. The gradients were fractionated into 0.35–ml fractions. The gradient fractions were analyzed for protein content, ATP synthase activity, and polypeptide profile on SDS-polyacrylamide gels (Supplementary Fig. 24b).

Purification of pyruvate carboxylase, an enzyme for carbon fixation. The pyruvate carboxylase gene was amplified from Bacillus subtilis 168 genomic DNA and cloned into a pTrc99a vector (AP Biotech Co., Argentina). This construct was kindly provided by P Kim (Catholic University of Korea, Korea). The pTrc99a vector containing pyruvate carboxylase was used to transform...
Escherichia coli strain BL21 (Novagen, Darmstadt, Germany) for expression and purification. The transformant was grown in lysogeny broth medium in the presence of ampicillin (50 µg/ml) (Duchefa Biochemie B.V., Netherlands) at 37 °C. Overnight cultures were diluted 1:100 and grown to produce an absorbance of 0.4 OD units at 600 nm at 37 °C. Cells were induced with 0.1 mM IPTG and grown for another 4 h at 37 °C. Cells were collected by centrifugation at 2,750g for 15 min at 4 °C. After cells were disrupted by sonication, unlysed cells and cell debris were removed by centrifugation at 4,000g for 20 min at 4 °C. The soluble fraction was obtained, and pyruvate carboxylase was purified using Ni-nitrotriacetic acid (NTA)–agaroose beads (Qiagen, Venlo, Netherlands). Finally, purified pyruvate carboxylase was subjected to SDS-PAGE to check its approximate purity (Supplementary Fig. 24c).

Preparation of small unilamellar vesicles, proteoliposomes, and photosynthetic artificial organelles. Small unilamellar vesicles (SUVs) were prepared from POPC, POPE, POPG, and cholesterol at a molar ratio of 2:1:1:1, and 0.25 mol% of NBD-PE or 0.25 mol% of Rhod-PE was added for visualization. Lipids were mixed in chloroform in a glass test tube, and the thin film obtained was kept in nitrogen gas and then under vacuum for more than 1 h to remove organic solvent residue. Lipids were hydrated in a defined volume of desired buffer solution (0.2 M sucrose, 2 mM HEPES (pH 8.0), and 2 mM MgSO₄) to yield a total lipid concentration of 5 mg/ml. This was then vortexed at a temperature above that of phase transition for 1 h. Frozen and thawed SUVs were obtained by repeating the following cycle 5 times: freezing the vesicles in liquid nitrogen and thawing the sample in a water bath at 40 ± 1 °C. The extrusion was performed by a mini-extruder (Avanti Polar Lipids, Alabaster, AL). The lipid solution was extruded with 100-nm polycarbonate membrane filters (Whatman, Inc., Newton, MA) and passed through the membrane filter 11 times, and the size distribution of vesicles was measured by dynamic light scattering.

Proteoliposomes containing photoconverters (PSII and PR) were reconstituted into SUVs by a detergent-mediated method 34. SUVs were solubilized into mixed micelles at detergent/lipid ratios of 1.5 (w/w) with DDM, and the solubilized photoconverters (PSII and PR) were added to the detergent–lipid mixture. After a 30-min incubation for micellar equilibration, the DDM was removed by a slow procedure consisting of four sequential additions of 30 mg Bio-Beads SM-2 per milligram of DDM, and the addition of chlorophyll-a in PSII was calculated by applying an equation: molecules = 8.70 × 10⁷/absorbance of 0.4 OD units at 600 nm at 37 °C. Cells were induced with 0.1 mM IPTG and grown for another 4 h at 37 °C. Cells were collected by centrifugation at 2,750g for 15 min at 4 °C. After cells were disrupted by sonication, unlysed cells and cell debris were removed by centrifugation at 4,000g for 20 min at 4 °C. The soluble fraction was obtained, and pyruvate carboxylase was purified using Ni-nitrotriacetic acid (NTA)–agaroose beads (Qiagen, Venlo, Netherlands). Finally, purified pyruvate carboxylase was subjected to SDS-PAGE to check its approximate purity (Supplementary Fig. 24c).

Preparation of giant unilamellar vesicles and giant proteoliposomes. Giant unilamellar vesicles (GUVs) were synthesized by the electroformation method 40. For calibration of intravascular pH, GUVs were prepared from a 1 mg/ml solution of lipids in chloroform and POPC, POPE, POPG, and cholesterol at a molar ratio of 2:1:1:1 and an additional 0.25 mol% of NBD-PE for visualization. 200 µl of the lipid solution was deposited on the conductive side of two-sided indium-tin oxide–coated glass slides (9 cm² area, 30–50Ω) with nitrogen gas. Organic solvents on the glass slides were removed by evaporation by placing the slides in a vacuum desiccator for 1 h. After drying, electroformation chambers were constructed by positioning 3-mm silicon spacers between two lipid-coated indium-tin oxide–coated glass slides. The slides were rehydrated with 0.2 M sucrose, 2 µM SNARF-1 (a pH-sensitive dye), and 2 mM HEPES at a pH of 6–9, and a 10-Hz sine wave of 1.2 V was applied across the chamber for 4 h using an arbitrary function generator (Tektronix Inc., OR). To detach the GUVs, a 4-Hz sine wave of 2 V was applied to the lipid-coated slides for 30 min. GUVs were then harvested using a wide-bore pipette tip and diluted 1:100 in resuspension buffer solution (0.2 M glucose, 2 mM MgSO₄, and 2 mM HEPES (pH 8.0)) and then a 10-Hz sine wave of 1.2 V was applied across the chamber for 8 h using an arbitrary function generator. To detach the GUVs, a 4-Hz sine wave of 2 V was applied to the lipid-coated slides for 30 min. GUVs were then harvested using a wide-bore pipette tip, diluted 1:100 in resuspension buffer solution (0.2 M glucose, 2 mM MgSO₄, 0.1 mM phenyl-p-benzoquinone, and 2 mM HEPES (pH 8.0)) and allowed to settle for 30 min before use. Giant proteoliposomes were imaged in confocal dishes that had been blocked with 1% Pluronic F-127 (Sigma–Aldrich Co., St. Louis, MO) for 1 h before use. All giant proteoliposomes were imaged within 4 h of electroformation.

Preparation of protocellular system. Giant unilamellar vesicles (GUVs) were prepared as above with DOPC, POPE, DOPS, and cholesterol at a molar ratio of 2:1:2:1, 1 mol% PEG2000-PE to prevent nonspecific interactions, 0.25 mol% of Rhod-PE for visualization, 1 mol% calcium ionophore A23187 (Sigma–Aldrich Co., St. Louis, MO), and 1 mol% magnesium ionophore I (Sigma–Aldrich Co., St. Louis, MO). The lipids were rehydrated with 2 mg/ml of photosynthetic organelles, 0.2 M sucrose, 20 mM Tris-HCl (pH 7.8), 0.1 mg/ml G-actin, 5 mM ADP, 3 mM NaH₂PO₄, 3 mM KH₂PO₄, 0.2 mM CaCl₂, and 0.1 µM Acti-stain 670 phalloidin for actin filament staining. Electroformation causes membrane swelling, which allows encapsulation of artificial organelles (100 nm) and other biological molecules (Supplementary Fig. 15a,b). To investigate membrane–actin interactions, GUVs were prepared from a liquid-disordered phase mixture (Ld) and from two different liquid-ordered phase mixtures (Lo1 and Lo2). Ld GUVs were prepared with DOPC, DOPS, and cholesterol at a molar ratio of 2:1:1. Lo1 GUVs were prepared with SM and cholesterol at a molar ratio of 1:1. Lo2 GUVs were prepared with SM and cholesterol at a molar ratio of 1:1 and 3 mol% PEG2000-PE to prevent membrane–actin interactions. To enable visualization, 0.25 mol% of Rhod-PE was added. Lipid hydration and electroformation were performed as above. For phase-separated protocellular system 43, phase-separated GUVs (Ld and Lo1) were prepared with a 2:1:3:1.5 molar ratio of DOPC:DOPS:SM: cholesterol incorporating 0.25 mol% of TopFluor-Chol or 0.25 mol% of Rhod-PE. An additional 3 mol% polysaturated PEG2000-PE was added to Ld and Lo2 phase-separated GUVs. Lipid hydration and electroformation were performed as above.

Confocal microscopy. Confocal microscopy was conducted using a Leica SP5 confocal microscope (Leica–Microsystems, Mannheim, Germany) equipped with ultraviolet and visible laser lines. GUVs, giant proteoliposomes, and...
SNARF-1 were excited at 488 and 532 nm at 10% laser power, and a pinhole was set at 1 or 3 units using a 63× oil immersion lens. SNARF-1 fluorescence emissions were captured in three channels at 520 ± 20 nm, 570 ± 20 nm, and 630 ± 20 nm. As SNARF-1 has been shown previously to be responsive to temperature, temperature was maintained between 22 °C and 24 °C during GUV observation. The use of 532 nm as the excitation wavelength minimized chlorophyll and other autofluorescence, as 532 nm lies outside of the absorption spectra of chlorophyll and autofluorescence.

For imaging protocol systems, a Zeiss LSM 710 confocal laser-scanning fluorescence microscope (Carl Zeiss GmbH, Jena, Germany) equipped with a C-Achromat 40×/1.2 water immersion objective was used in multitrack mode. NBD-PE and TopFluor-Chol samples were excited at 488 nm using an argon laser line with a 488/543/633 beam splitter and imaged at 493–563 nm on track 1. Rhod-PE samples were excited at 543 nm using an HeNe laser line with a 488/543/633 beam splitter and imaged at 563–628 nm on track 2. Actin-stain 670 phallolidin samples were excited at 633 nm using an HeNE laser line with a 488/543/633 beam splitter and imaged at 638–755 nm on track 2. All images and three-dimensional reconstructions of images were analyzed and processed using ImageJ-processing software (ImageJ, NIH, Bethesda, MD) and Zeiss Zen 2012 (Carl Zeiss GmbH, Jena, Germany). To minimize any side effects from the laser source, the cells were imaged at 1.97 fps, and it was confirmed that the confocal microscope’s scanning laser sources did not influence pH during imaging (Supplementary Fig. 4a).

Calibration of intravesicular pH. Intravesicular pH was measured by ratio metric analysis of SNARF-1 fluorescence.45 GUVs were prepared by electroformation with 2 μM SNARF-1 encapsulated in buffer solution (0.2 M sucrose, 2 mM HEPES) at pH 6–9. GUVs were then harvested using a wide-bore pipette tip and diluted 1:200 in resuspension buffer solution (0.2 M glucose, 2 mM HEPES) to dilute the external SNARF-1 dye. GUVs containing SNARF-1 were imaged with a confocal microscope with an HeNe laser source tuned to 543 nm. The ratio of fluorescence intensity of SNARF-1 was determined from a 1 mg/ml solution of lipids in chloroform. To assist visualization of GUVs by light microscopy, we added 0.25 mol% of NBD-PE to the lipid solution. Giant vesicles with a diameter of 15–30 μm were collected for use in calibrating intravesicular pH.

Measuring the photocurrents of PSI and PR. The photocurrents of PSI, PR, and both photoconverters in the bilayer were measured with the Port-a-Patch automated patch clamp system (Nanoion Technologies GmbH, Munich, Germany). We used giant proteoliposomes i.e., GUVs reconstituted with PSI, PR, or both photoconverters) to form a bilayer44,47 in an aperture (approximately 2 μm in diameter) of a patch clamp chip (NPC-1, Nanoion Technologies GmbH, Munich, Germany).

For the formation of a planar lipid bilayer containing PR, 10 μl of the proteoliposome-containing solution was pipetted onto a patch clamp micro-con structed chip. Giant proteoliposomes were positioned on the aperture of the micro-constructed chip by applying negative pressure. Generally, –5 to ~20 mbar was sufficient for reliable positioning within a few seconds after giant proteoliposome addition. When the giant proteoliposomes touched the glass surface of the chip, they burst and formed planar bilayers with a seal resistance of at least 1 GΩ. Membrane capacitance could be estimated on the order of a few femtofarads. Owing to the small volumes required, rapid perfusion was a necessity. Photocurrents were induced by illumination by a SPECTRA light engine (Lumencor, Beaverton, OR) at 1.14 mW/cm², which was filtered by blue (438 ± 24 nm), green (540 ± 27 nm), or red (660 ± 52 nm) band-pass filters. Light was directed onto the lipid bilayer via an optical fiber (Doric Lens Inc., Quebec, Canada). All experiments were carried out at room temperature.

Measurement of light-induced intravesicular pH. The intravesicular pH of giant proteoliposomes was measured by ratiometric analysis of SNARF-1 fluorescence.45 Choosing to measure single giant proteoliposomes minimized the side effects of reducing the unincorporated vesicles and residues. Giant proteoliposomes were prepared through dehydration of a solution of small proteoliposomes.41,42 Small proteoliposomes were prepared as described above and diluted to 1 mg/ml. 200 μl of the solution was dehydrated on the conductive side of two-sided indium-tin oxide–coated slides (9 cm² area, 30–50Ω) using a vacuum desiccator at 4 °C for 3 h. The lipids were rehydrated with 0.2 M sucrose, 2 μM SNARF-1, 2 mM HEPES (pH 8.0), and 2 mM MgSO₄, and then a 10-Hz sine wave of 1.2 V was applied across the chamber for 8 h using an arbitrary function generator. GUVs were then harvested using a wide-bore pipette tip and diluted 1:100 in resuspension buffer solution (0.2 M glucose, 2 mM MgSO₄, 0.1 mM phenyl-p-benzoquinone, and 2 mM HEPES (pH 8.0)) and allowed to settle for 30 min before use. The giant proteoliposomes were illuminated with a fiber-optic illuminator with a halogen bulb equipped with heat-absorbing filters (Fiber Lite PL-800, Dolan-Jenner Industries, Lawrence, MA) that emitted broadband white light or light that was filtered by blue (438 ± 24 nm), green (540 ± 27 nm), or red (660 ± 52 nm) band-pass filters at 1.14 mW/cm² by controlling a neutral density filter (NDL-25C-4, Thorlabs, Newton, NJ) with a power meter (PM160, Thorlabs, Newton, NJ). The giant proteoliposomes were illuminated every 2 min, and the light was turned off during imaging.

Quantification of ATP synthesis by artificial organelles. ATP synthesis by artificial organelles was determined with the luciferin–luciferase system in bulk-vesicle solution using an ATP Determination Kit (Molecular Probes, Eugene, OR) and following the manufacturer’s protocol. The procedure was adapted from a paper by Andrea Hazard et al.48. Artificial organelles containing PR, PSI, and ATP synthase were prepared as described above. The artificial organelles were suspended in a solution containing 20 mM Tris-HCl (pH 8.0), 3 mM KH₂PO₄, 3 mM NaH₂PO₄, 5 mM NaHCO₃, 5 mM ADP, 2 mM MgCl₂, and reaction buffer, 1 μM FCCP (a protonophore) and 1 μM oligomycin (an inhibitor of ATP synthase) were added as the negative control. The stirred suspension was illuminated with a fiber-optic illuminator with a halogen bulb equipped with heat-absorbing filters (Fiber Lite PL-800; Dolan-Jenner Industries, Lawrence, MA) that emitted broadband white light or light that was filtered by blue (438 ± 24 nm), green (540 ± 27 nm), or red (660 ± 52 nm) band-pass filters at 1.14 mW/cm² by controlling a neutral density filter (NDL-25C-4, Thorlabs, Newton, NJ) with a power meter. Multiple samples in a quartz cuvette were illuminated with various light wavelengths and for various durations (1–33 min, 1-min intervals). Samples were analyzed by measuring fluorescence emission at 560 nm using an F-7000 FL Spectrophotometer (Hitachi, Tokyo, Japan).

Quantification of carbon fixation of artificial organelles. To determine the level of oxaloacetate (OAA), an OAA fluorometric assay kit (BioVision Inc., CA) and the gas chromatography/mass spectrometry (GC/MS) method were used. The artificial organelles were suspended in a solution containing 20 mM Tris-HCl (pH 8.0), 3 mM KH₂PO₄, 3 mM NaH₂PO₄, 5 mM NaHCO₃, 5 mM ADP, 2 mM MgCl₂, 5 μM pyruvate, 0.5 μM pyruvate carboxylase, 0.1 μM acetyl-coenzyme A (acetyl-CoA), and reaction buffer. The resulting suspension was mixed with buffer mix solution containing the assay kit’s enzyme, developer, and probe. 2 mM phenylacetic acid (inhibitor of pyruvate carboxylase) were added as a negative control. The stirred suspension was illuminated with a fiber-optic illuminator with a halogen bulb equipped with heat-absorbing filters (Fiber Lite PL-800; Dolan-Jenner Industries, Lawrence, MA) that emitted broadband white light or light that was filtered by blue (438 ± 24 nm), green (540 ± 27 nm), or red (660 ± 52 nm) band-pass filters at 1.14 mW/cm² by controlling a neutral density filter (NDL-25C-4, Thorlabs, Newton, NJ) with a power meter. Multiple samples in a quartz cuvette were illuminated with various light wavelengths and for various durations (1–20 min, 1-min intervals). Samples were analyzed by measuring fluorescence emission using an F-7000 FL Spectrophotometer (Hitachi, Tokyo, Japan).

For GC/MS analysis, artificial organelles containing PSI-PR photoconverters that were illuminated for 1, 5, 10, 15, and 20 min were transferred to a 3,000 MW Amicon ultra filter device (Millipore Inc., Billerica, MA) to collect reactants and then centrifuged at 4,000g for 30 min. The pellets were taken and dried in a vacuum concentrator at room temperature. For the derivatization of pyruvate and OAA, dried residues were resuspended in 25 μl MTBSTFA + 1% TBDMS and 25 μl of acetonitrile and then incubated for 1 h at 60 °C. Silylated derivatives of pyruvate and OAA were separated on a DB-5 capillary column (30 m × 0.250 mm, 0.25 μm; Agilent Technologies, Foster City, CA). Analyses
were performed with an Agilent 6890 gas chromatograph coupled to an Agilent 5973 mass-selective detector (MSD) with the following chromatographic conditions: initial temperature 50 °C for 2 min, increasing to 150 °C at 20 °C/min over 5 min and finally to 300 °C at 8 °C/min for 20 min. The front inlet temperature was 250 °C operating with a split ratio of 1:10. The MSD ion source and interface temperature were 280 °C. Helium was used as carrier gas (1.0 mL/min). The MSD operated in EI mode at 70 eV. A scan mode of 15–650 m/z was used for the analyses. GC/MS data were acquired and processed using Agilent MassHunter Workstation software.

**Polymerization of actin in photosynthetic protocellular systems.** A suspension of protocellular systems was transferred to confocal dishes that had been blocked with 1% Pluronic F-127 (Sigma–Aldrich Co., St. Louis, MO) for 1 h before use to prevent nonspecific interactions between the bottom of the glass and actin. The suspension of protocellular systems was diluted 1:10 in resuspension buffer solution (0.2 M glucose, 20 mM Tris-HCl (pH 7.8), 0.5 mM CaCl₂, and 0.5 mM MgCl₂). After incubation at room temperature for 30 min to allow ions to be transported into the cells, actin polymerization was induced in the protocellular systems by illuminating them with a fiber-optic illuminator that emitted either broadband white light at 1.14 mW/cm² or light that was filtered by green (540 ± 27 nm) or red (660 ± 52 nm) band-pass filters at 1.14 mW/cm² by controlling a neutral density filter (NDL-25C-4, Thorlabs, Newton, NJ) with a power meter (PM160, Thorlabs, Newton, NJ). The samples in confocal dishes were illuminated using light at various wavelengths and for various durations. To minimize any side effects from the laser source, the cells were imaged at 1.97 fps and the light was turned off during the measurements.

**Calculating the area of the actin sphere and the contour curvature of protocellular systems.** The area of the actin sphere was calculated using the following equation:

\[ A = \pi \times \left( \frac{x}{2} \right)^2 \]

where all derivatives are with respect to t. The first and last two pixels of each digital curve were ignored to avoid end effects. The probability distribution of contour curvatures was calculated as a histogram with fixed bin widths and bin positions for each time point (Supplementary Note 2).

**Statistics.** Data are presented as mean ± s.d. The significance between data sets was measured using two-tailed unpaired Student’s t-test. P values were not considered significant when \( P > 0.05 \) and significant when \( P < 0.05 \) and \( P < 0.01 \).

**Code availability.** Custom code for calculating contour curvature of protocellular system organelle membrane systems is supplied in Supplementary Note 2.

**Life Sciences Reporting Summary.** Further information on experimental design is available in the Nature Research Reporting Summary linked to this article.

**Data availability.** The authors confirm that the data supporting the findings of this study are available within the paper and its supplementary information files. Additional source data are available from the corresponding author upon reasonable request.

## Experimental design

1. **Sample size**
   - Describe how sample size was determined.

   **No statistical methods were used to predetermine sample size**

2. **Data exclusions**
   - Describe any data exclusions.

   **No data were excluded for proteoliposome and artificial organelles experiments. For encapsulated organelle membrane system, it is excluded that the encapsulation does not proceed stochastically (See Supplementary Figure 16b).**

3. **Replication**
   - Describe whether the experimental findings were reliably reproduced.

   **No attempts at replication failed.**

4. **Randomization**
   - Describe how samples/organisms/participants were allocated into experimental groups.

   **For purification of membrane proteins, a random selection of spinach and protein expressed e.coli was chosen. For proteoliposomes and artificial orangelles, there was no biased use of particular sample tubes or chemical reagents. Though across batches, individuals were randomly processed. No other randomization is applicable for this study.**

5. **Blinding**
   - Describe whether the investigators were blinded to group allocation during data collection and/or analysis.

   **Not applicable. Group allocation was not a feature for experiments and associated data for in-vitro experiments.**

Note: all studies involving animals and/or human research participants must disclose whether blinding and randomization were used.

6. **Statistical parameters**
   - For all figures and tables that use statistical methods, confirm that the following items are present in relevant figure legends (or in the Methods section if additional space is needed).

   - n/a
   - Confirmed

   **The exact sample size (n) for each experimental group/condition, given as a discrete number and unit of measurement (animals, litters, cultures, etc.)**

   **A description of how samples were collected, noting whether measurements were taken from distinct samples or whether the same sample was measured repeatedly**

   **A statement indicating how many times each experiment was replicated**

   **The statistical test(s) used and whether they are one- or two-sided (note: only common tests should be described solely by name; more complex techniques should be described in the Methods section)**

   **A description of any assumptions or corrections, such as an adjustment for multiple comparisons**

   **The test results (e.g. P values) given as exact values whenever possible and with confidence intervals noted**

   **A clear description of statistics including central tendency (e.g. median, mean) and variation (e.g. standard deviation, interquartile range)**

   **Clearly defined error bars**

*See the web collection on statistics for biologists for further resources and guidance.*

Nature Biotechnology: doi:10.1038/nbt.4140
Software

Policy information about availability of computer code

7. Software

Describe the software used to analyze the data in this study.

Student’s unpaired t-tests (Microsoft Excel 2016 MSO Version 16.0.9029.2167 64bit), Confocal image (Carl zeiss ZEN 2.3 SP1 black 64bit 14.0.0.201), 3D reconstruction of confocal images, (Image.J 1.51n), Curvature calculation (MATLAB 9.1.0.441655 R2016b), Curve fitting (OriginPro 8 SR0 v8.0724 B724)

For manuscripts utilizing custom algorithms or software that are central to the paper but not yet described in the published literature, software must be made available to editors and reviewers upon request. We strongly encourage code deposition in a community repository (e.g. GitHub). Nature Methods guidance for providing algorithms and software for publication provides further information on this topic.

Materials and reagents

Policy information about availability of materials

8. Materials availability

Indicate whether there are restrictions on availability of unique materials or if these materials are only available for distribution by a for-profit company.

No restrictions.

9. Antibodies

Describe the antibodies used and how they were validated for use in the system under study (i.e. assay and species).

No Antibodies were used.

10. Eukaryotic cell lines

a. State the source of each eukaryotic cell line used.

No Eukaryotic cell lines were used in this study.

b. Describe the method of cell line authentication used.

No Eukaryotic cell lines were used in this study.

c. Report whether the cell lines were tested for mycoplasma contamination.

No Eukaryotic cell lines were used in this study.

d. If any of the cell lines used are listed in the database of commonly misidentified cell lines maintained by ICLAC, provide a scientific rationale for their use.

No Eukaryotic cell lines were used in this study.

Animals and human research participants

Policy information about studies involving animals; when reporting animal research, follow the ARRIVE guidelines

11. Description of research animals

Provide details on animals and/or animal-derived materials used in the study.

No animals were used in this study

Policy information about studies involving human research participants

12. Description of human research participants

Describe the covariate-relevant population characteristics of the human research participants.

No human research participants were used in this study
Supplementary Figure 1

Overview of steps in the construction of photosynthetic protocellular systems

(a) The small unilamellar vesicles were made with phospholipids. (b) Three types of small proteoliposomes were created by reconstituting PSII, PR, and PSII-PR into the vesicles. (c) Artificial organelles were made by reconstituting ATP synthases into proteoliposomes. (d) The encapsulated organelle system was built by encapsulating organelles and reactants in a giant unilamellar vesicle (10–100 μm diameter) containing membrane-bound ion transporters (magnesium ionophores). (e) Information of ionic solution and lipid compositions in protocellular system was summarized.
Supplementary Figure 2

Estimated numbers of photoconverters reconstituted into the phospholipid vesicle and bilayer

(a) Ratios of phospholipids to photoconverters in proteoliposomes were calculated based on the absorption spectrum and absorption coefficients of photoconverters (see Methods). (b–d) Number of chromophores in a giant proteoliposome (b), a lipid bilayer patch (c), and a small proteoliposome (d). See Methods for calculations. Error bars indicate mean ± s.d. of seven independent experiments.
Supplementary Figure 3

Calibration of a ratiometric SNARF-1 indicator for intravesicular pH measurement

(a) Fluorescence images of SNARF-1 encapsulated GUVs captured at 500–540 nm (green: membrane), 560–600 nm (orange: acidic forms of SNARF-1), and 610–650 nm (red: basic forms of SNARF-1). The images are representative of twelve independent experiments. (b) Fluorescence emission spectrum of SNARF-1 in a buffered solution (2 mM HEPES, 0.2 M sucrose) at various pH values. The data are representative of three independent experiments with similar results. (c) Ratios of the acidic to the basic forms of SNARF-1 in GUVs at various intravesicular pH values. The blue line indicates the sigmoidal fitting function of the relation between the intravesicular pH level and the SNARF-1 intensity ratio, where the inflection point at pH = 7.22 corresponds to the pKa indicator. Error bars indicate mean ± s.d. of twelve independent experiments. All scale bars = 20 μm.
Supplementary Figure 4 | See next page for caption
Supplementary Figure 4

Intravesicular pH measurement of giant proteoliposomes by the fluorescence ratiometric method

(a–d) Images and fluorescence intensity ratios of giant proteoliposomes reconstituted in the absence of photoconverters (a) and in the presence of PR (b), PSII (c), and PR-PSII (d). All scale bars = 20 μm. Data are presented as box plots showing the mean as well as the 25th and 75th percentiles. End of whiskers are set at 1.5 × s.d. above and below third and first quartiles. The images are representative of independent experiments (Control: n = 15, PR: n = 7, PSII: n = 5, PSII-PR: n = 8 biologically independent experiments). (e) Intravesicular pH levels of giant proteoliposomes converted from fluorescence intensity ratios (a–d) using a calibration curve (Supplementary Fig. 3c). Following 5 min of illumination, the intravesicular pH of PSII, PR, and PSII-PR co-reconstituted giant proteoliposomes significantly decreased compared with the pH levels measured in the absence of photoconverters. In contrast, pH levels were maintained in narrow ranges following 5 min in the light-off condition. Error bars and colored bands indicate mean ± s.d. of independent experiments (Control: n = 15, PR: n = 7, PSII: n = 5, PSII-PR: n = 8 biologically independent experiments).
Supplementary Figure 5

Effect of POPG on the activity of photoconverters (PSII and PR) in proteoliposomes

(a) Comparison of intravesicular levels of PSII and PR reconstituted proteoliposomes in the presence and absence of POPG. (b) The presence of POPG significantly improved the intravesicular pH decrease rates of PSII (2.9×) and PR (2.4×) reconstituted proteoliposomes. (a, b) Error bars and colored bands indicate mean ± s.d. of seven independent experiments. (b) (b) *$P = 5.78 	imes 10^{-9}$, **$P = 6.20 	imes 10^{-9}$, $P$ values were calculated by Student's two-sided t test.
Supplementary Figure 6

GC/MS analysis of reaction products by artificial organelles

(a) Chromatograms of artificial organelles after illumination for 1, 5, 10, 15, and 20 min, derivatized with MTBSTFA + 1% TBDMCS (see Methods). (b) Mass spectra of PA derivative (retention time = 12.1 min) (upper panel) and OAA derivative (retention time = 22.3 min) (lower panel) were derivatized with MTBSTFA + 1% TBDMCS. The spectra are consistent with pyruvate and OAA, with NIST Mass Spectral Library matches (PA: 91% OAA: 99%). Consistent fragment ions for the TBDMCS derivatives of pyruvate and OAA were observed at m/z 139 and 417, respectively. (a, b) One representative data from two independent experiments are shown.
Supplementary Figure 7

ATP synthesis coupled with carbon fixation

(a, b) The graphs show the relation between ATP and OAA production. The graphs were acquired from PSII and PR reconstituted and PR-PSII co-reconstituted artificial organelles during 5 (a) and 20 (b) min of white light illumination (see Supplementary Discussion). (c) The three plots overlap, which suggests that OAA production is directly correlated to ATP production. (a–c) Error bars and the colored bands indicate mean ± s.d. of ten independent experiments. (b) P values were calculated by Student’s two-sided t-test.

Nature Biotechnology: doi:10.1038/nbt.4140
Supplementary Figure 8

Wavelength- and pH-dependent activities of PSII and PR reconstituted lipid bilayers

(a, b) Photocurrent traces of PR or PSII reconstituted lipid bilayers at pH 8.0 (a) and pH 6.0 (b) under white (400–700 nm), blue (438 ± 24 nm), green (540 ± 27 nm), and red (660 ± 27 nm) light illumination. Data are representative of seven independent experiments. Error bars indicate mean ± s.d. of seven independent experiments. c, d, Summary of wavelength- and pH-dependent activities of PSII and PR at pH 8.0 (c) and pH 6.5 (d). At pH 8.0, both PR and PSII generated negative photocurrents, indicating that light stimulation increased proton concentration inside the lipid bilayer. At pH 6.5, PR and PSII generated positive and negative currents, respectively, showing that PR exhibits pH-dependent bidirectional activity under green light illumination at pH 6.5 and pH 8.0 and that PR and PSII can be independently activated by green and red light, respectively, at pH 6.5.
Supplementary Figure 9

Wavelength- and pH-dependent photocurrents of PS II and PR in PSII-PR co-reconstituted proteoliposomes

(a) At pH 8.0, both green and red light illumination of PSII-PR co-reconstituted proteoliposomes generated a negative photocurrent. Data are representative of ten independent experiments. Error bars indicate mean ± s.d. of ten independent experiments. (b) At pH 6.5, green light illumination changed the photocurrent direction, indicating that the photocurrent direction of PSII-PR co-reconstituted proteoliposomes can be controlled by red and green light at pH 6.5.
Supplementary Figure 10

Wavelength- and pH-dependent intravesicular pH of PSII and PR reconstituted giant proteoliposomes

(a, b) Intravesicular pH levels (a) and rate of decrease in intravesicular pH of PSII (left) and PR (right) reconstituted giant proteoliposomes (b) at the same initial intravesicular pH level (pH 8) under illumination with light of various wavelengths (white [400–700 nm], blue [438 ± 24 nm], green [540 ± 27 nm], red [660 ± 27 nm]). The changes in intravesicular pH between PSII and PR reconstituted proteoliposomes were significantly different under green and red light illumination, indicating that PSII and PR can be independently activated by red and green light, respectively (n = 7 biologically independent experiments). (c, d) Intravesicular pH levels (c) and intravesicular pH decrease rates (d) of PSII (left) and PR (right) reconstituted giant proteoliposomes under white light illumination at various initial pH levels. The intravesicular pH level of a PSII reconstituted proteoliposome gradually decreases as pH decreases at pH 6–9 because PSII activity decreases at lower pH levels. In contrast, the intravesicular pH level of a PR reconstituted proteoliposome is saturated at pH 7.5 because PR has a pH-dependent bidirectional proton-pumping ability, pumping in at pH > 8 and pumping out at pH < 7.5 (n = 8 biologically independent experiments). (a–d) Error bars and colored bands indicate mean ± s.d. of independent experiments. (b) *P = 0.026, **P = 6.351 × 10⁻⁸, ***P = 0.0002, ****P = 0.0006. P values were calculated by Student’s two-sided t test.
Supplementary Figure 11

Wavelength-dependent ATP synthesis of artificial organelles

PSII and PR reconstituted and PSII-PR co-reconstituted organelles synthesized ATP upon light simulation at 1.14 mW/cm². The amount of ATP synthesized varied with the light wavelength: blue (438 ± 24 nm), green (540 ± 27 nm), and red (660 ± 27 nm). The integration of FCCP, a protonophore, dramatically decreased ATP synthesis, demonstrating that ATP synthesis is coupled with the proton motive force generated by the organelles. ATP concentration was measured with luciferin/luciferase (n = 10 biologically independent experiments). Colored bands indicate mean ± s.d. of ten independent experiments.
**Supplementary Figure 12**

**Optimization of PSII, PR, and ATP synthase ratios in PSII-PR co-reconstituted organelles**

**a.** Intravesicular pH level of PSII-PR co-reconstituted organelles with various ratios of PSII to PR under white light illumination. The curve representing proteoliposomes with a PSII:PR ratio of 1:2 shows an intermediate pH level between that measured for PR (0:1 ratio) and PSII (1:0 ratio) reconstituted proteoliposomes (n = 7 biologically independent experiments), indicating that the contribution of PR and PSII to the intravesicular pH of PSII-PR co-reconstituted proteoliposomes is similar. 

**b, c.** Amounts of ATP synthesized by the artificial photosynthetic organelles with various ratios of ATP synthase to lipid. As the ratio of ATP synthase to lipid increased, the ATP synthesis rate increased, until it saturated at 1:2,000. There is no statistical difference between the 1:2,000 and 1:1,000 ratios (n = 5 biologically independent experiments). Error bars and colored bands indicate mean ± s.d. of independent experiments. P values were calculated by Student’s two-sided t test.
ATP synthesis of the artificial organelles impeded by the light-off condition

The artificial organelles slowly stopped ATP synthesis when white light was turned off after 15 min of illumination (n = 10 biologically independent experiments), compared to ATP synthesis following green light illumination (Fig. 3e, right). The photosynthetic organelles continued ATP synthesis despite initiation of the light-off condition owing to residual PMF, which led to a slower quenching rate compared with the rate observed with green light illumination. The gray band indicates mean ± s.d. of independent experiments.
Supplementary Figure 14

Carbon fixation coupled with ATP synthesis by artificial organelles

The amount of oxaloacetate (C4) synthesized from pyruvate (C3) increased when the photosynthetic organelles were stimulated with white, red, and green light (n = 10 biologically independent experiments). Each colored band indicates mean ± s.d. of ten independent experiments.
Supplementary Figure 15

Fabrication process for protocellular systems capable of actin polymerization when optically stimulated

Electroformation causes lipid membrane swelling (1), which allows encapsulation of artificial organelles and other biological molecules with buffer solution (0.2 M sucrose, 20 mM Tris-HCl [pH 7.8], 0.1 mg/ml G-actin, 5 mM ADP, 3 mM NaH₂PO₄, 3 mM KH₂PO₄, 0.2 mM CaCl₂, and 0.1 μM Acti-stain™ 670 phalloidin for actin filament staining) (2) in a GUV. Ionophores then begin transporting ions (e.g., magnesium ions or calcium ions) for ion-dependent reactions. Light stimulation causes the encapsulated organelles to generate ATP, which is used in cascading reactions that lead to actin polymerization (4).
Supplementary Figure 16

Investigation of encapsulation and lipid fusion between the outer membrane (protocellular systems) and artificial organelles

(a) Fluorescence image of the outer membrane (red) with various sizes (50, 100, 200, and 1,000 nm) of artificial organelles (green) after electroformation. Encapsulation of organelles and empty membrane system are indicated by # and O, respectively. The images are representative of three independent experiments. All scale bars = 50 μm. (b) The rate of intracellular encapsulation decreases with increasing organelle size. Error bars indicate mean ± s.d. of three independent experiments. (c) Fluorescence image of the outer membrane (red) and organelles (green) with different charge combinations and sizes. Fusion of the outer membrane and the organelles was interrupted by electrostatic repulsive interactions with negatively charged lipids as well as phosphatidylglycerol (PG) and phosphatidylserine (PS).
The positively charged lipid (1,2-dioleoyl-3-trimethylammonium-propane [TAP]) induced membrane fusion. The results of membrane fusion clearly are indicated by fluorescent signal on the outer membrane. The images are representative of four independent experiments with similar results. All scale bars = 20 μm.
Supplementary Figure 17

Effect of a higher concentration of magnesium ions on the thickness of actin filaments in protocellular systems

(a–c) Protocellular systems were prepared with 0.5 mM Mg²⁺. The final concentration of magnesium ions was adjusted to 2 mM. Actin filament thickness clearly increased (c). Three-dimensional reconstructions of a partial protocellular system allow investigation of actin thickness via examination of a cross-section of the membrane system (b, d). The images are representative of three independent experiments with similar results. All scale bars = 20 μm.
Supplementary Figure 18

Relationship of ATP synthesis in PSII-PR co-reconstituted organelles over time

The red dashed line shows the results for an allometric equation on a log-log plot. The inset indicates the linear axes. Colored curves indicate mean ± s.d. of measurements seven independent experiments.

\[
\text{Time (t)} = a^b \\
a = 69.17 \\
b = 0.50
\]
Investigation of actin polymerization in protocellular systems under a range of conditions

(a) Protocellular systems (red), artificial organelles (green), and G-actin (white) were exposed to white light for 2 hr. (b) Positive control: Encapsulated ATPs allowed for actin polymerization in protocellular systems. (c-g) Negative controls: Protocellular systems were not able to initiate actin polymerization in the absence of light simulation (c), artificial organelles (d), ADP (e), magnesium ions (f), or G-actin (g), indicating that actin polymerization is completely dependent on ATP synthesis by artificial organelles in the membrane system. All scale bars = 20 μm. Red, green, and white colors indicate the outer membranes of protocellular systems, artificial organelles, and actin, respectively. (a–g) The images are representative of three independent experiments with similar results.
Supplementary Figure 20

Light-induced deformation of protocellular systems made of a single-phase membrane mixture: liquid-disordered (Ld) or liquid-ordered (Lo1, Lo2) phase membrane mixtures

(a-c) Single-phase membrane systems were prepared with an Ld phase membrane mixture (DOPC, DOPE, and DOPS) (a), an Lo1 phase membrane mixture (SM and cholesterol) (b), or an Lo2 phase membrane mixture (SM, cholesterol, and polysaturated-PEG2000-PE) (c). As actin polymerization proceeded membrane–actin interaction altered cell shape. The membrane system made with the Ld mixture became wrinkled owing to attractive membrane–actin interactions, whereas the membrane system made with the Lo1 or Lo2 mixture was spherical owing to repulsive membrane–actin interactions. The gap caused by the strong repulsive membrane–actin interaction, induced by the presence of PEG2000 in the lipid membrane, may be clearly observed in (c). (a–c) The images are representative of three independent experiments with similar results. All scale bars = 20 μm.
Supplementary Figure 21

Characteristic shapes of phase-separated protocellular systems made of various lipid combinations

(a) An Ld-Lo2 mixture led to polarization of actin filaments in the membrane system. (b) An Ld-Lo mixture led to mushroom-shaped membrane systems owing to weak actin–Lo interactions. Green, red, and white colors indicate organelles, the outer membrane, and actin filaments, respectively. All scale bars = 20 μm. (a, b) The images are representative of five independent experiments with similar results.
Supplementary Figure 22

Light-induced deformation of phase-separated protocellular systems made of Ld and Lo2 phase membrane mixtures

(a) Images and membrane curvature maps of phase-separated protocellular systems made using the phase separation method (see Methods). In the phase-separated membrane system, the Ld phase membrane (red dotted line, top left) was segregated from the Lo2 phase membrane (blue dotted line, bottom right). Polymerized actins strongly attached to the Ld phase membrane owing to strong membrane–actin interactions but not to the Lo2 phase membrane owing to repulsive membrane–actin interactions. The difference between these local membrane–actin interactions induced changes in the global shape of the membrane system as actin polymerization proceeded: As the F-actin sphere grew, the global curvature of the Ld part of the membrane increased but that of the Lo2 part decreased. (b, c) Magnified curvature maps of regions in (a) indicated by dotted lines, created following 120 min (b) and 210 min (c) of light illumination. The local curvature of the Ld phase membrane side (b and c, left column) significantly changed owing to the strong attractive interaction, compared the local curvature of the Lo2 side (b and c, right column) during actin polymerization. (a–c) The images and data are representative of five independent experiments with similar results. All scale bars = 20 μm.

Nature Biotechnology: doi:10.1038/nbt.4140
Isolation and characterization of PSII complexes.

(a) Sucrose gradient of solubilized membranes, showing eight green bands. (b) SDS-PAGE analysis of the sucrose gradient fractions. (c) Absorption spectrum of each band, normalized to the maximum in the red region. (a–c) The images and data are representative of three independent experiments with similar results.
Supplementary Figure 24

SDS-PAGE analysis.

(a) ATP synthase. (b) Proteorhodopsin. (c) Pyruvate carboxylase expression. (a–c) The images are representative of two independent experiments with similar results.